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Acid and Base Worksheet - Answers. 1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a)  $\text{HNO}_3 + \text{OH}^- \rightarrow \text{H}_2\text{O} + \text{NO}_3^-$ .  $\text{HNO}_3$  and

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NO<sub>3</sub><sup>-</sup> make one pair OH<sup>-</sup> and H<sub>2</sub>O make the other. b) CH<sub>3</sub>NH<sub>2</sub> + H<sub>2</sub>O ( CH<sub>3</sub>NH<sub>3</sub><sup>+</sup> + OH<sup>-</sup>)

## **Acid and Base Worksheet - Answers**

Introduction. Originally the terms acid and base referred to taste. The practice of classifying substances according to their acidic (sour) or basic (alkaline or bitter) properties dates back to ancient times. An acid was something with a sour taste, such as lemon juice, and a base was something with a bitter taste, such as tonic water.

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## **Acids and Bases Worksheet Middle School - DSoftSchools**

10. Explain in terms of the partial charge on hydrogen why NaOH is a base, HClO is a weak acid and HClO<sub>4</sub> is a strong acid. 11. Why is HCl a strong acid and HClO a weak acid? 12. Why are HCl

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and HClO<sub>4</sub> both strong acids? 13. For each of the reactions below, classify the reactants as an acid or a base and the products as the conjugate acid or ...

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Naming Acids And Bases Worksheet Answer Key. 31 scaffolded questions that start relatively easy and end with some real challenges. 1 Acids, Bases, and pH. 2 Acid Base Reactions. A difference in 1 unit is a 10X difference in concentration For example; a liquid with pH of 3 is 10X more acidic than a liquid with a pH of 4. ...

### **Unit 5 Worksheets Acids And Bases Answer Key**

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when an alkali is added to an acid?, ...

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Short Answer type Question 10 Give a reason why (a) Great care should be taken while handling laboratory acids and bases. (b) Solid powder of baking soda does not turn dry litmus paper to blue in color. Question 11 Define acid rain and how it is produced? What are its harmful effect? Question 12 Define neutralization reaction with an example.

## **Worksheet for Class 7 Science Chapter 5: Acid Base and Salt**

Acid Nomenclature Worksheet KEY Write the formula for each of the acids listed below: 1. Nitric acid  $\text{HNO}_3$  2. Chloric acid  $\text{HClO}_3$  3. Acetic acid  $\text{HC}_2\text{H}_3\text{O}_2$  4. Hydrobromic acid  $\text{HBr}$  5. Sulfurous acid  $\text{H}_2\text{SO}_3$  6. Chlorous acid  $\text{HClO}_2$  7. Hydrochloric acid  $\text{HCl}$  8.

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Phosphoric acid  $H_3PO_4$ . Nitrous acid  $HNO_2$ .

## Acid Nomenclature Worksheet Name

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## Acids and Bases Worksheet Answers | Homeschooldressage.com

Acids and Bases Theories and Neutralization Practice Q Answer Key 1) List four properties of an acid. Dissociates into  $H^+$  ions in solution, turns litmus paper red, pH is less than 7, conducts electricity, caustic and corrosive, tastes sour, reacts with metals to produce hydrogen gas through single replacement.

## Piersa, Amanda / Unit 11: Acids and Bases

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Acid and Base Worksheet 1) Using your knowledge of the Brønsted-Lowry theory of acids and bases, write equations for the following acid-base reactions and indicate each conjugate acid-base pair: a)  $\text{HNO}_3 + \text{OH}^-$  b)  $\text{CH}_3\text{NH}_2 + \text{H}_2\text{O}$  c)  $\text{OH}^- + \text{HPO}_4^{2-}$  2) The compound NaOH is a base by all three of the theories we discussed in class.

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