

## Atomic Structure Practice 1 Answers

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### Atomic Structure Practice 1 Answers

Atomic Structure Practice Answers. Q1. i) K<sup>+</sup> protons = 19; neutrons = 20; electrons = 18 (1+ charge, so 1 electron removed) ii) H<sup>+</sup> protons = 1; neutrons = 0; electrons = 0 (1+ charge, so 1 electron removed) iii) H<sup>-</sup> protons = 1; neutrons = 0; electrons = 2 (1- charge so 1 electron added) Q2. i) <sup>37</sup>Cl<sup>-</sup> atomic number of 17 may also be given ii) <sup>24</sup>Mg<sup>2+</sup> atomic number of 12 may also be given.

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Basic Atomic Structure Worksheet H and the 1. 2. 4. 5. 6. 8. 9. The 3 particles of the atom are: protons (p<sup>+</sup>) neutrons (n<sup>0</sup>) electrons (e<sup>-</sup>) Their respective charges are: The number of protons in one atom of an element determines the atom's number Of electrons determines the of the element in one atom of an element. It also in a neutral atom of that element. The atomic number

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### Atomic Structure Practice 1 Worksheet Answers

Atoms consist of protons, neutrons, and electrons. If the mass of neutrons and electrons were made half and two times respectively to their actual masses, then the atomic mass of <sup>6</sup>C<sup>12</sup>. a) Will remain approximately the same. b) Will become approximately two times. c) Will remain approximately half.

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Atomic Structure Quiz Answers. 1. All elements are composed of. atoms; compounds; neutrons; molecules; 2. What is the mass number of carbon (C)? 12; 6; 13; C; 3. How many electrons are in an atom of carbon (C)? (Remember APE! Atomic Number = # of Protons and # of Electrons) 6; 13; 12; C; 4. How many electrons are in an atom of carbon (C)? (Remember APE!

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