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Question: 1.11 - Name the parameters that characterise a unit cell. Answer:

Unit cells are characterized on six parameters - dimensions along three edges and three angles between their edges, i.e. a , b , c which are edges and α , β and γ which are angles between the

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edges. Question: 1.12 - Distinguish between (i) Hexagonal and monoclinic unit ...

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4.1. For the reaction $R \rightarrow P$, the concentration of a reactant changes

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from 0.03 M to 0.02 M in 25 minutes.
Calculate the average rate of reaction using units of time both in minutes and seconds.

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some solid compounds which are referred to as 12-16 compounds. For example, ZnS, CdS, CdSe, HgTe, etc. In these compounds, the bonds have ionic character. More Resources for CBSE Class 12:

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3.1. How would you determine the standard electrode potential of the system $Mg^{2+} | Mg$? Ans: A cell will be set up consisting of $Mg/MgSO_4 (1 M)$ as one electrode and standard hydrogen electrode $Pt, H_2 (1 atm) | H^+ (1 M)$ as

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second electrode, measure the EMF of the cell and also note the direction of deflection in the voltmeter.

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supplied at the end. They provide a useful study tool for students. For instructors, they are a good resource for homework, quiz, or test questions, or practice for the AP Chemistry test.

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Chemistry Class 12 Chemistry. NCERT TEXTBOOK QUESTIONS SOLVED. 11.1 Classify the following as primary, secondary and tertiary alcohols. Ans. Primary alcohols: (i), (ii), (iii) Secondary alcohols: (iv), (v)

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Question 14: Suggest two materials other than hydrogen that can be used as fuels in fuel cells. Answer: Methane and Methanol. Question 15: Explain how rusting of iron is envisaged as setting up of an electrochemical cell. Answer: The chemistry of corrosion is quite complex but it can be considered as an electrochemical phenomenon. Oxidation

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