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Chemistry Matter Change Chapter 12

Chemistry: Matter and Change Chapter 12 Vocabulary. STUDY. PLAY. kinetic-molecular theory. the behavior of matter in terms of particles in motion. elastic collision. collision in which no kinetic energy is lost. temperature. measure of the average kinetic energy of the particles in a sample of matter.

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Section 12.1 Gases Section 12.2 Forces of Attraction Section 12.3 Liquids and Solids Section 12.4 Phase Changes Exit CHAPTER States of Matter 12 Click a hyperlink to view the

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corresponding slides. •Use the kinetic-molecular theory to explain the

Chemistry: Matter and Change

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240 Chemistry: Matter and Change • Chapter 12 Solutions Manual CHAPTER 12 SOLUTIONS MANUAL 30. Compare and contrast sublimation and evaporation. In both processes, the substances become a vapor. During sublimation, the substance goes from the solid phase directly to the vapor phase. During evaporation, particles in a liquid gain enough

States of Matter

Where To Download Chemistry Matter And Change Chapter 12 4 Study Guide Answers in which some properties of the material change, but the identity of the matter does not. When we heat the liquid water, it changes to water vapor. But even though the physical properties have changed, the molecules are exactly the same as before.

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Chemistry: Matter and Change Teacher Guide and Answers 7 Study Guide - Chapter 12 - States of Matter Section 12.1 Gases 1. motion 2. a. small b. forces c. random d. elastic; kinetic 3. KE

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1/2 mv² 4. Temperature 5. true 6. true 7. false 8. true 9. true 10. false 11. true 12. false 13. a 14. a 15. d 16. d 17. b 18. b 19. b 20. barometer 21. Evangelista Torricelli

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Supplemental Problems Chemistry: Matter and Change • Chapter 2 1 Data Analysis Data Analysis 1. A sample of aluminum is placed in a 25-mL graduated cylinder containing 10.0 mL of water. The level of water rises to 18.0 mL. Aluminum has a density of 2.7 g/mL. Calculate the mass of the sample. 2. Saturn is about 1 429 000 km from the Sun.

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