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term or phrase that best completes each statement or best answers each question. ____ 1. According to the kinetic-molecular theory, particles of an ideal gas a. attract each other but do not collide.

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CHAPTER 10 REVIEW States of Matter
SECTION 3 SHORT ANSWER Answer the following questions in the space provided. 1. Match description on the right to the correct crystal type on the left. b ionic crystal (a) has mobile electrons in the crystal c covalent molecular crystal (b) is hard, brittle, and nonconducting a metallic crystal (c)

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1 typically has the lowest melting point of the four

10 States of Matter - Ms. Agostine's Chemistry Page

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ANSWER Write the answers to the following questions in the space provided. 22.

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CHAPTER 9 REVIEW Stoichiometry

MIXED REVIEW SHORT ANSWER Answer

the following questions in the space
provided. 1. Given the following

equation: $C_3H_4(g) + xO_2(g) \rightarrow 3CO$

$2(g) + 2H_2O(g)$ 4 a. What is the value of

the coefficient x in this equation? 40.07

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g/mol b. What is the molar mass of C_3H_4 ? 2 mol O_2 :1 mol H_2O c. What is the mole ratio ...

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MIXED REVIEW SHORT ANSWER Answer
the following questions in the space

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provided. 1. Classify each of the following as a homogeneous or heterogeneous substance.

homogeneous a. sugar homogeneous d. plastic wrap homogeneous b. iron filings heterogeneous e. cement sidewalk heterogeneous c. granola bar 2. For each type of investigation, select the most appropriate ...

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